

Available online at www.sciencedirect.com

Tetrahedron Letters

Tetrahedron Letters 48 (2007) 5411–5413

Synthesis of an octahydroindolinone scaffold for a diversity-based chemical compound library

Gregg F. Keaney* and Charles W. Johannes

Infinity Pharmaceuticals, Inc., 780 Memorial Drive, Cambridge, MA 02139, USA

Received 16 May 2007; accepted 4 June 2007 Available online 9 June 2007

Abstract—The synthesis of a chemical compound library using diversity-oriented synthesis (DOS) is discussed. The library is structurally inspired by the *Amaryllidaceae* alkaloids, a family of natural products which has been known to demonstrate potent antiviral and antineoplastic activity. Highlights of this work include the rapid, high-yielding construction of the octahydroindolinone core and the solid-phase diversification of the lactam using a neutral phosphazene base. © 2007 Elsevier Ltd. All rights reserved.

The octahydroindolinone ring system is a structural motif which is frequently observed in the Amaryllidaceae family of natural products. These natural products have demonstrated a wide range of biological activities including antiviral, antitumor, acetylcholinesterase inhibitory, immunostimulatory, and antimalarial activi-ties.^{[1](#page-1-0)} Representative members of the Amaryllidaceae family, mesembrine (1) and obliquine (2), are shown in Figure 1.

In pursuit of synthesizing novel chemical structures using our diversity-oriented synthesis (DOS) platform, we became interested in the octahydroindolinone framework as a scaffold for a chemical compound library. Using a chemistry-driven approach, we aimed to develop an efficient, solution-phase route to an octahydroindolinone core, which contained three diversity sites that

Figure 1. Octahydroindolinone ring system and Amaryllidaceae natural products.

0040-4039/\$ - see front matter © 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2007.06.018

could be elaborated on solid-phase into a diverse, multi-thousand member compound library.

Recently, Padwa et al. have developed an elegant method to access the hexahydroindolinone framework via an oxabicyclic intermediate using an intramolecular Diels– Alder furan/ring-opening sequence. $2-8$ This oxabicycle, rich in stereochemical complexity and exhibiting an array of versatile functional groups, was recognized as a potential starting point for a diversity-based chemical compound library.

In accordance with a literature procedure, acylation oftert-butyl N-(2-furyl)carbamate (3) provided an intermediary imidofuran which underwent spontaneous [4+2] cyclization at room temperature to provide 7-oxabicy $clo[2.2.1]$ heptene 4 [\(Scheme 1](#page-1-0)).^{[6](#page-1-0)} Initial efforts to install diversity were focused on opening the oxa-bridge in an S_N^2 fashion using nitrogen (sulfonamides, anilines, imides, amines, etc.) and oxygen (primary alcohols and phenols) nucleophiles in the presence of Rh^I catalyst.^{[9,10](#page-2-0)} However, in all cases, the starting material was recovered unchanged, perhaps due to reversible addition of the nucleophile. Alternatively, it was found that treatment of 4 with acetone or *para*-anisaldehyde in the presence of tin chloride cleanly provided the corresponding acetals 5 and 6 in excellent yields.^{[11](#page-2-0)} Irreversible trapping of the oxonium ion by the newly generated alkoxide likely accounts for the high regio- and diastereoselectiv-ity was observed in the isolated products.^{[12](#page-2-0)}

Due to substrate compatibility issues encountered while advancing the para-methoxybenzylidene acetal series,

^{*} Corresponding author. Tel.: +1 617 453 1128; fax: +1 617 453 1001; e-mail: gregg.keaney@ipi.com

Scheme 1. Synthetic route to the core.

attention was focused on advancing acetonide 5. To this end, hydrogenation of the vinyl imide of 5 led to exclusive formation of the *cis*-ring fused product.^{[13](#page-2-0)} Next, selective deprotection of the acetonide in the presence of the Boc group furnished diol 8. Since one of the hydroxyl groups of 8 would serve as a handle for loading to solid support, differentiation of the two alcohols was required at this stage. Ultimately, it was found that catalytic dibutyltin oxide could beemployed to facilitate a regioselectivenosylation to provide nosylate 9 in 76% yield.¹⁴⁻¹⁶ To our knowledge, regioselective introduction of the nosyl (Ns, para-nitrobenzenesulfonyl) protecting group via the stannylene acetal has not been previously reported in the literature. Deprotection of the Boc group with magnesium perchlorate completed the synthesis of the octahydroindolinone core 10, the structure of which was confirmed by single crystal X-ray analysis (see Fig. 2). $6,17-19$

In preparation for solid-phase diversification, the octahydroindolinone core 10 was loaded onto polystyrene Lantern^{TM} (Mimotopes L-series) with a loading level of 14.5 μ mol/Lantern[™]. As shown in Scheme 2, cleavage of the nosyl group using the Fukuyama protocol revealed secondary alcohol 12, which was poised for sub-sequent diversification.^{[20,21](#page-2-0)} Additionally, lactam 11 was functionalized with a series of activated electrophiles using BTPP (tert-butylimino-tri(pyrrolidino)phosphorane), a neutral phosphazene base recently employed by Shaw et al. in a similar transformation.²

In summary, we have developed a six-step route to a novel octahydroindolinone core in 16% overall yield.

Figure 2. ORTEP diagram of compound 10.

Scheme 2. Preliminary solid-phase diversification.

The solution-phase sequence requires only two chromatographic purifications (intermediates 4 and 9) and has been scaled to provide multi-gram quantities of 10. Initial solid-phase chemistry involving nosyl deprotection and lactam functionalization has enabled entry into two solid-phase diversification pathways. Further solidphase elaboration of this core and additional work on related scaffolds will be reported in due course.

Acknowledgments

The authors thank Dr. Lisa A. Marcaurelle, Dr. Marta Nevalainen, and Dr. Vesa Nevalainen for their contributions to this work.

References and notes

- 1. Jin, Z. Nat. Prod. Rep. 2003, 20, 606–614.
- 2. Padwa, A.; Dimitroff, M.; Waterson, A. G.; Wu, T. J. Org. Chem. 1997, 62, 4088–4096.
- 3. Padwa, A.; Brodney, M.; Dimitroff, M.; Liu, B.; Wu, T. J. Org. Chem. 2001, 66, 3119–3128.
- 4. Padwa, A.; Lee, H. I.; Rashatasakhon, P.; Rose, M. J. Org. Chem. 2004, 69, 8209–8218.
- 5. Padwa, A.; Rashatasakhon, P.; Ozdemir, A. D.; Willis, J. J. Org. Chem. 2005, 70, 519–528.
- 6. Padwa, A.; Wang, Q. Org. Lett. 2004, 6, 2189–2192.
- 7. Padwa, A.; Wang, Q. J. Org. Chem. 2006, 71, 7391–7402.
- 8. Padwa, A.; Wang, Q. J. Org. Chem. 2006, 71, 3210–3220.
- 9. Lautens, M.; Dockendorff, C.; Fagnou, K.; Malicki, A. Org. Lett. 2002, 4, 1311–1314.
- 10. Lautens, M.; Fagnou, K.; Yang, D. J. Am. Chem. Soc. 2003, 125, 14884–14892.
- 11. Acetonide formation $(4 \rightarrow 5)$ using these conditions was reported concurrently by Padwa et al. (Ref. 7).
- 12. This reaction may proceed through an epoxide intermediate; see: Vyvyan, J. R.; Meyer, J. A.; Meyer, K. D. J. Org. Chem. 2003, 68, 9144–9147.
- 13. A recent publication, which describes reduction attempts on a related hexahydroindolinone system, demonstrates that the steric bulk of the substituent at the ring junction affects the stereochemical outcome of the reduction: see: Brodney, M. A.; Cole, M. L.; Freemont, J. A.; Kyi, S.; Junk, P. C.; Padwa, A.; Riches, A. G.; Ryan, J. H. Tetrahedron Lett. 2007, 48, 1939–1943.
- 14. Martinelli, M. J.; Vaidyanathan, R.; Pawlak, J. M.; Nayyar, N. K.; Dhokte, U. P.; Koecke, C. W.; Zollars, L. M. H.; Hoher, E. D.; Khau, V. V.; Kosmrlj, B. J. Am. Chem. Soc. 2002, 124, 3578–3585.
- 15. David, S.; Hanessian, S. Tetrahedron 1985, 41, 643–663.
- 16. Interestingly, nosylation did not proceed without dibutyltin oxide in the reaction mixture.
- 17. Stafford, J. A.; Brackeen, M. F.; Karanewsky, D. S.; Valvano, N. L. Tetrahedron Lett. 1993, 34, 7873–7876.
- 18. Characterization data for 10: White solid (mp 147– 148 °C); ¹H NMR (400 MHz, DMSO- d_6) δ 8.46 (d, $J = 9.2$ Hz, 2H), 8.22 (d, $J = 9.2$ Hz, 2H), 4.76 (m, 1H),

3.92 (t, $J = 4.2$ Hz, 1H), 3.68 (s, 3H), 3.63 (td, $J = 3.4$, 10.2 Hz, 1H), 2.61 (d, $J = 16.2$ Hz, 1H), 2.30 (d, $J = 16.2$ Hz, 1H), 2.20 (td, $J = 4.5$, 15.5 Hz, 1H), 1.99 (ddd, $J = 3.5$, 5.2, 15.5 Hz, 1H), 1.83 (comp m, 2H) ppm; 13 C NMR (100 MHz, DMSO-d₆) δ 174.9, 174.0, 151.3, 142.9, 130.4, 127.9, 125.4, 124.3, 82.3, 65.5, 53.4, 52.4, 49.1, 42.7, 33.5, 30.7 ppm; IR (HATR FT-IR) 3416 (br m), 3232 (w), 2948 (m), 1695 (s), 1532 (s), 1350 (m), 1189 (m), 903 (m), 620 (m) cm^{-1} ; HRMS (FAB) m/z found: 432.1070 [calcd for $C_{16}H_{20}N_2O_{10}S$ (M+H₂O)⁺: 432.0839]; Anal. Calcd (C₁₆H₁₈N₂O₉S): C, 45.46; H, 4.56; N, 6.51; S, 7.01.

- 19. Crystal data for 10: $C_{16}H_{18}N_2O_9S$, $M = 414.38$, $T =$ 193(2) K, triclinic, space group P-1, $Z = 2$, $a =$ 8.1740(11), $b = 10.9303(15)$, $c = 11.9711(17)$ \AA , $\alpha =$ 113.496(2)°, $\beta = 95.460(2)$ °, $\gamma = 107.828(2)$ °, $V = 904.4(2)$ Å³, $D_x = 1.522$ Mg/m³, $R_1 = 0.0475$, $wR_2 =$ 113.496 $(2)^{\circ}_{2}$, 904.4(2) Å³, 0.1062. Crystallographic data for 10 have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 647518. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK, (fax: +44-(0)1223-336033 or e-mail: deposit@ccdc.cam.ac.uk).
- 20. Fukuyama, T.; Jow, C.-K.; Cheung, M. Tetrahedron Lett. 1995, 36, 6373–6374.
- 21. Maligres, P. E.; See, M. M.; Askin, D.; Reider, P. J. Tetrahedron Lett. 1997, 38, 5253–5256.
- 22. Mitchell, J. M.; Shaw, J. T. Angew. Chem., Int. Ed. 2006, 45, 1722–1726.